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Process-dependent residual trapping of CO₂ in sandstone

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Abstract This paper demonstrates that the nature and extent of residual CO₂ trapping depend on the process by which the CO₂ phase is introduced into the rock. We compare residual trapping of CO₂ in Berea Sandstone by imbibing water into a core containing either exsolved CO₂ or CO₂ introduced by drainage. X-ray computed tomography measurements are used to map the spatial distribution of CO₂ preimbibition and postimbibition. Unlike during drainage where the CO₂ distribution is strongly influenced by the heterogeneity of the rock, the distribution of exsolved CO₂ is comparatively uniform. Postimbibition, the CO₂ distribution retained the essential features for both the exsolved and drainage cases, but twice as much residual trapping is observed for exsolved CO₂ even with similar preimbibition gas saturations. Residually trapped exsolved gas also disproportionately reduced water relative permeability. Development of process-dependent parameterization will help better manage subsurface flow processes and unlock benefits from gas exsolution.

1. Introduction

Geological storage projects would introduce millions of tons of CO₂ into deep underground formations resulting in complex flow and transport processes operating over millennial timescales [Metz *et al.*, 2005]. Initially, flow and transport of CO₂ will be dominated by multiphase displacements of in situ fluids [Pruess and Garcia, 2002; Nordbotten *et al.*, 2005]. Over time and with distance from the injection well, a progressively larger fraction of the CO₂ will dissolve into reservoir brines due to postinjection gravitational equilibration [Hesse *et al.*, 2008; Doughty and Pruess, 2004] and convective mixing [Riaz *et al.*, 2006]. Up to 5% mass fraction of CO₂ can dissolve into reservoir brines depending on pressure, temperature, and brine composition [Duan and Sun, 2003]. Exsolution of dissolved CO₂ occurs when pore pressures decline [Enouy *et al.*, 2011; Zuo *et al.*, 2012, 2013] or temperatures rise [Luhmann *et al.*, 2012], resulting in the formation of a separate CO₂ phase. For example, if CO₂-saturated brine migrates upward through faults, leaking boreholes or seals, a separate phase of CO₂ exsolves as the pressure decreases [Falta *et al.*, 2013; Oldenburg *et al.*, 2012]. Exsolution can also occur during CO₂-enhanced oil recovery, particularly near the production wells [Alizadeh *et al.*, 2011] and can improve secondary recovery when exsolved CO₂ blocks water flow paths, displacing water into oil-filled pores and mobilizes residual oil [Zuo and Benson, 2013].

Numerical simulations using discretized mass and energy balance equations are commonly used to predict the influence of these complex processes on the long-term fate of CO₂ or on the ultimate recovery from an oilfield [Oldenburg *et al.*, 2001; Kumar *et al.*, 2004]. Key inputs include relative permeability and capillary pressure characteristic curves and trapping coefficients. These characteristic curves are typically treated as single-valued functions of the saturation, or in some cases include hysteresis, or history-dependent values [Juanes *et al.*, 2006; Doughty, 2007; Altundas *et al.*, 2011]. Implicitly, these models assume that characteristic curves are not a function of the process by which a secondary phase is introduced into the rock, for example, by drainage in the cases of CO₂ injection or by exsolution, in the case of declining pressures.

Differences in the spatial distribution of CO₂ are expected to result from the process by which separate phase CO₂ is introduced into the rock. For example, during drainage, subcore-scale heterogeneity has a strong influence in the spatial distribution and transport of CO₂ [Chaouche *et al.*, 1994; Perrin and Benson, 2010; Shi *et al.*, 2011]. Capillary heterogeneity creates barriers that prevent CO₂ from entering portions of the rock with high capillary entry pressure [Ferrand and Celia, 1992; Ringrose *et al.*, 1993; Illangasekare *et al.*, 1995; Krevor *et al.*, 2011; Shi *et al.*, 2011; Krause *et al.*, 2011]. After the initial pore invasion when saturations are no longer changing, the drainage-induced gas distribution behind the invasion front is at local

capillary equilibrium [Leverett, 1941; Lenormand et al., 1983; Blunt et al., 2002], although nonequilibrium capillarity effects are observed during initial invasion [Barenblatt et al., 2003; Berg et al., 2013]. Imbibition following drainage will displace some of the CO₂ but trap the rest as disconnected ganglia that show strong dependence on the topology of the nonwetting fluid, wettability, and pore geometries [Chaudhary et al., 2013; Joekar-Niasar et al., 2013; Kim et al., 2012; Tanino and Blunt, 2012; Tokunaga et al., 2013]. Pore-scale imaging of postdrainage trapping shows that trapped supercritical CO₂ in a water-wet system yields a power law cluster size distribution [Andrew et al., 2013], as predicted from percolation theory [Lorenz and Ziff, 1998], and capillary heterogeneity has been shown to increase trapping [Krevor et al., 2011; Saadatpoor et al., 2010].

In contrast to drainage, an exsolved CO₂ phase consists primarily of disconnected rather than pore-spanning bubbles that are uniformly distributed throughout the rock and not strongly influenced by capillary heterogeneity [Luhmann et al., 2012; Zuo et al., 2013]. The capillary pressure in each bubble is determined by local pore geometry and consequently can vary throughout the rock. Liberation of gas has been found to significantly reduce both gas and liquid mobility in gas-oil systems [Fishlock et al., 1988; Grattoni et al., 1998; Naylor et al., 2000; Tang and Firoozabadi, 2003] and in CO₂-water system [Zuo et al., 2012] under high pressures while Fry et al. [1997], Zhao and Ioannidis [2011], and Enouy et al. [2011] show that water during exsolution at near-ambient conditions is as mobile as during drainage, and exsolved air/CO₂ is mobilized by capillary instability and buoyancy forces. So far, there has been no explanation for this discrepancy. Nucleation and postnucleation processes that control the distribution and morphology of exsolved gas in the pore space depend on rock types, fluid properties, and pressure conditions [Li and Yortsos, 1995]. In light of the disconnected nature of the gas phase following exsolution [Zuo et al., 2013], we hypothesize that under reservoir conditions, postexsolution residual gas trapping will be different than postdrainage trapping and require process-specific parameterization of the residual trapping coefficient and relative permeability.

In this paper, we compare postdrainage and postexsolution CO₂/water trapping under the same pressure and temperature conditions. Preimbibition and postimbibition gas distributions are compared and quantitatively analyzed. Relative permeabilities at imbibition endpoints are also measured to demonstrate the process dependency and influence of gas morphology on multiphase flow properties.

2. Experimental Methods

2.1. Materials and Equipment

For the experiments we use a core of Berea Sandstone obtained from a commercial quarry (Cleveland Quarries), 15 cm in length and 5 cm in diameter. The core is heated at 700°C overnight to stabilize swelling clays before the experiments. The core-flooding apparatus is similar to the setup described by Perrin and Benson [2010]. Five syringe pumps (Teledyne Isco, model 500D and 1000D) are used to circulate fluids at elevated pressure and temperature in a closed loop. The core is placed in an aluminum core holder, and an X-ray computed tomography scanner (General Electric HiSpeed CT/i) is used to obtain rock porosity and phase saturations (500 μm pixel resolution in cross sections; 1 mm interval in length) during the experiments [Akin and Kovscek, 2003]. Deionized water and CO₂ are used as wetting and nonwetting phases.

2.2. Procedure

The rock is scanned sequentially while air filled (atmosphere pressure), CO₂ filled (4.14 MPa), water filled (4.14 MPa), and equilibrated carbonated water filled (4.14 MPa) to calculate rock porosity (Figure 1a) and obtain background scans of the core. All reported X-ray CT data are the average of two replicate scans. Absolute permeability of the core, 507 mD, is measured when the core is water filled. The core and injected fluids are maintained at 50°C throughout experiments. A 2.76 MPa confining pressure is applied to the core.

The drainage process is started by coinjecting CO₂ at 9 mL/min and carbonated water at 1 mL/min into the core at 4.14 MPa. Multiple scans are taken to measure the CO₂ saturation distribution and determine when it stabilizes. CO₂ injection is stopped when the CO₂ saturation distribution stabilizes. An imbibition process follows immediately by injecting carbonated water at 1 mL/min. Multiple scans are taken until the residual gas saturation is reached. After the gas saturation stabilizes at 1 mL/min, the carbonated water injection rate

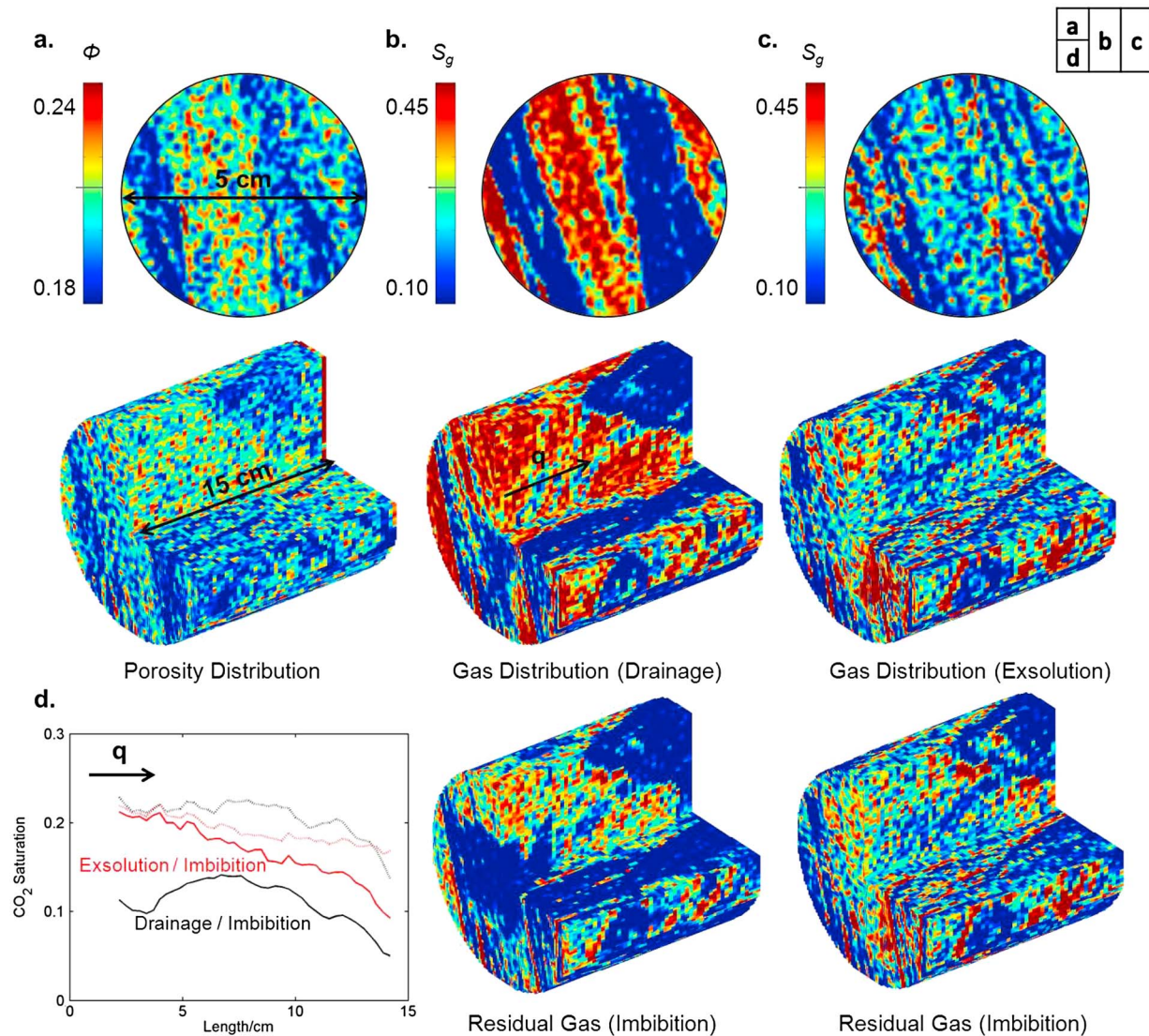


Figure 1. Porosity distributions of the (a) middle cross section and the whole core, (b) corresponding CO₂ saturation distributions in drainage/imbibition and (c) exsolution/imbibition, and (d) cross section CO₂ saturation profiles.

is then increased to 5 and then 15 mL/min. After the gas saturation stabilized at each flow rate, the relative permeability to the aqueous phase is measured.

For the exsolution process, the core is fully saturated with carbonated water equilibrated at 10.34 MPa. Afterward, the upstream end of the core is sealed, and the downstream is regulated at a constant pressure decline rate of 0.1 or 1 MPa/h until the pore pressure is reduced to 4.14 MPa and held constant thereafter at 50°C. The pressure/temperature conditions simulate an upward migration of carbonated brine from a designated storage reservoir around 1000 m to a shallower depth around 400 m which is typical for CO₂ storage projects. Small pressure gradients are observed across the core during depressurization, and this is similar to the condition when exsolution occurs in a reservoir, although the pressure decline rates are expedited. While rate-dependent behaviors of exsolved gas have been observed with hydrocarbons [Scherpenisse *et al.*, 1994], no significant difference in gas aggregation/trapping are observed in our experiments between 0.1 and 1 MPa/h pressure decline rates. Results discussed here are for a pressure decline rate of 1 MPa/h. Identical to the drainage/imbibition experiment, following exsolution, sequential imbibition at 1, 5, and 15 mL/min with carbonated water (equilibrated at 4.14 MP) injections are conducted. The injection rate is adjusted after the CO₂ saturation stabilizes.

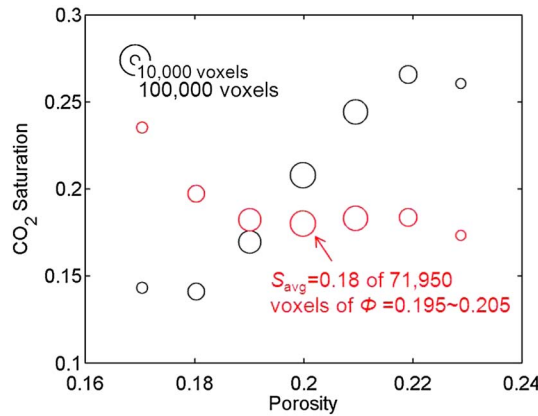


Figure 2. Porosity saturation correlations for drainage (black) and exsolution (red). Saturations are average values over voxels within porosity intervals (0.01 width). Circle area represents the number of voxels within the particular porosity interval.

nonwetting phase distribution in drainage. High-porosity regions have significantly higher CO₂ saturations than regions with just several percent lower porosity. In contrast with drainage, CO₂ exsolution is less affected by the capillary heterogeneity, and CO₂ is distributed more evenly, although the influence of bedding structures on the saturation distribution is still somewhat evident. Overall, the whole core has similar average CO₂ saturations in both cases, 0.20 and 0.18 for drainage and exsolution, respectively.

Figure 2 illustrates clear statistical differences between the saturation distribution for the exsolution and drainage cases. There is a positive correlation between drainage CO₂ saturation and porosity while the exsolution CO₂ saturation is independent of porosity. The area of circle represents the number of voxels of the whole core within a certain porosity interval. For example, the red circle centered at 0.20 porosity and 0.18 saturation shows that the average CO₂ saturation is 0.18 for all voxels with porosity 0.20 ± 0.005 (71,950 individual voxels). The standard deviation of each aggregation of drainage CO₂ saturation is ~14% and ~10% in exsolution CO₂ saturation. This large spread is primarily caused by small-scale heterogeneity and sedimentary structures.

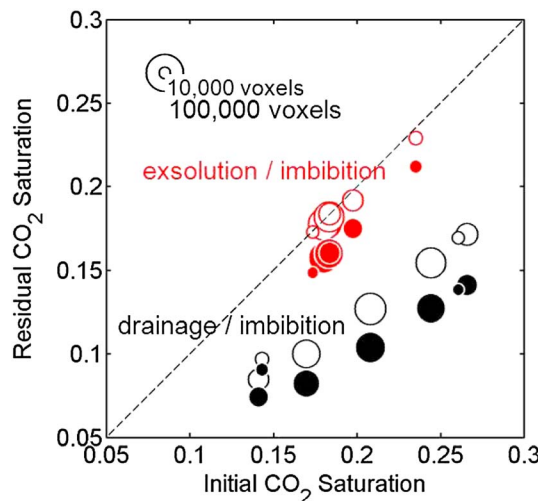


Figure 3. Residual CO₂ saturations versus the initial CO₂ saturations in drainage/imbibition (black) and exsolution/imbibition (red). Unit slope is plotted in dashed line. Residual CO₂ saturations after 1 and 15 mL/min carbonated water injection are in open and filled circles.

3. Results

Figure 1 shows the middle cross section and the whole core's porosity distributions (Figure 1a), CO₂ distributions in drainage/imbibition process (Figure 1b), and exsolution/imbibition process (Figure 1c). Cross-sectional averaged CO₂ saturation profiles are shown in Figure 1d (black dotted/solid curves: drainage/imbibition; red dotted/solid curves: exsolution/imbibition). Residual gas saturations for both scenarios shown are equilibrated at 15 mL/min carbonated water injection. The core has an average porosity of 0.20, and alternating low- and high-porosity zones appear as high-angle bedding planes along the length of the core. This feature affected the drainage CO₂ saturation distribution significantly and manifested the importance of small-scale capillary heterogeneity on

Figure 3 shows residual CO₂ saturations after 1 and 15 mL/min carbonated water injections versus the initial CO₂ saturations. Data points are the aggregation of voxels defined identically in Figure 2. Similar to previous studies [Pentland et al., 2011; Krevor et al., 2012], in a water-wet system, the residual CO₂ saturation increases monotonically with the initial CO₂ saturation in all four cases. However, capillary trapping is much more extensive with exsolved gas. This sole explanation for this difference must stem from the differences in gas morphology, because the interfacial tension between CO₂ and water is the same for both experiments. Figure 4 shows corresponding Land trapping coefficients, $C = \frac{1}{S_{gr}} - \frac{1}{S_{gi}}$, for postdrainage and postexsolution imbibition [Land, 1968], where S_{gi} and S_{gr} are initial and residual gas saturations. During carbonated water injection at 1 mL/min, almost all exsolved gas is immobile, and the curve

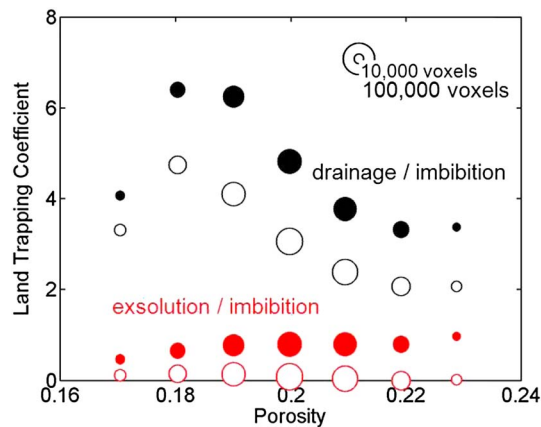


Figure 4. Land coefficients [Land, 1968] in drainage/imbibition (black) and exsolution/imbibition (red). Coefficients of 1 and 15 mL/min carbonated water injection are marked in open and filled circles.

trapping in high-porosity regions of the rock tested here. However, increased trapping in high-porosity regions could actually be the consequence of downstream capillary barriers [Krevor et al., 2011] and sedimentary structures which lead to bypass during imbibition [Huang et al., 1995].

4. Discussion

Injection of CO₂ at low capillary numbers (10⁻⁸–10⁻⁷) leads to strongly capillary heterogeneity controlled gas distribution during drainage. Capillary equilibrium contributes to the large gas saturation contrast observed (Figure 1b) between neighboring regions. On the contrary, gas exsolved almost uniformly in all porosity regions (Figure 2) which indicates this in situ gas evolution is not hindered by capillary heterogeneity, at least over the range of pore sizes relevant here (10–100 μm pore radius in Berea Sandstone).

During imbibition, almost all exsolved gas ($S_{gr} = 0.18$) is trapped, and only a small fraction is mobilized ($S_{gr} = 0.16$) when the water injection rate increases from 1 ($Ca \sim 10^{-7}$) to 15 ($Ca \sim 10^{-6}$) mL/min. In contrast, the postdrainage gas saturation decreases from 0.2 to 0.12 during 1 mL/min water injection and further down to 0.10 during 15 mL/min injection. These observations can be explained by the morphology of the exsolved CO₂. Bubbles nucleate instantaneously under large supersaturation [Firoozabadi and Kashchiev, 1996] and distribute uniformly in the core. Individual bubbles grow by diffusion from the aqueous phase into the gas phase. As the bubble grows, capillary pressure will increase when the gas-liquid interface penetrates into pore throats. When the capillary threshold at the throats is overcome, the bubble will expand into neighboring pores. Due to the high CO₂/water interfacial tension, the gas-liquid interface requires high energy to span multiple pores, and capillarity instability in high-aspect ratio pores leads to snap off [Roof, 1970]. A dispersed, residual phase-like morphology is formed as CO₂ evolves in porous media and becomes trapped in pores [Zuo et al., 2013].

The unique morphology of exsolved gas exhibits both opportunities and challenges in management of subsurface flow processes. In geological carbon storage, seals with low permeabilities and high capillary entry pressures are expected to prevent gas or supercritical phase CO₂ from leaking out of the storage reservoir. However, the seal does not prevent dissolved CO₂ from migrating upward. If a dissolved phase migrates into the seal and a gas phase exsolves due to depressurization, the mobility of the aqueous phase will be further reduced, thus suppressing further aqueous phase migration [Falta et al., 2013]. This CO₂ gas barrier can also be manipulated for water conformance and mobility control in oil/gas applications [Zuo et al., 2013]. By decreasing the reservoir pressure, CO₂ exsolves as a secondary residual phase in addition to trapped oil. A highly dispersed morphology and a strong nonwetting character of the CO₂ phase would promise an effective block of the water flow therefore improving sweep efficiency.

In the experiments, water relative permeability is measured at the end of imbibition process at 15 mL/min (~10 m/d), which is higher than normal flow rates under reservoir conditions. The water relative permeability

at the imbibition endpoint is: $k_{rw} = 0.28$ at $S_{gr} = 0.16$ in postexsolution imbibition and $k_{rw} = 0.65$ at $S_{gr} = 0.10$ in postdrainage imbibition, with similar initial CO_2 saturations (drainage: 0.20; exsolution: 0.18). This parameterization discrepancy, along with different trapping coefficients, is not reflected in reservoir simulators today. It is clear that not only history-dependent behavior (hysteresis) but also process-dependent behavior will result in very different multiphase flow parameters. Poorly interconnected exsolved gas interferes more strongly with the water flow than injected gas, even at the very high flow rates tested here. For the CO_2 -brine system, at high pressures and in consolidated rocks, using postdrainage imbibition relative permeability parameterizations for simulations involving exsolution will lead to overestimation of gas and liquid phase flows. The large mobility reduction of the CO_2 and water observed here is consistent with results of exsolution associated with drainage in different sandstones studied previously [Zuo *et al.*, 2012].

Consequently, the challenge that lies ahead is the need to distinguish and account for process-specific parameterizations of key inputs such as relative permeability and trapping coefficients. For a simulation involving only one of the processes, the solution can be straightforward, as long as the appropriate process-specific parameterization is included. However, in general, it is possible that multiple processes are acting concurrently in different portions of the subsurface, and multiple parameterizations are required. Tracking both history-dependent and process-dependent phenomenon will be required for accurate simulation. Development of process-dependent parameterization will help better manage subsurface flow processes and unlock benefits from gas exsolution.

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